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DIFFUSION BONDING OF RAFM STEELS: EVOLUTION OF INTERFACIAL OXIDE LAYER WITH PRESSURE AND MICROSTRUCTURE AND MECHANICAL PROPERTY AFTER POST BONDING HEAT TREATMENT

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Abstract

The effect of increasing the bonding pressure from 10 to 20 MPa (5 MPa interval) at 1050 °C for 60 minutes on the diffusion bonded joint of reduced activation ferritic/martensitic (RAFM) steel was investigated. The results indicate that the quality of the bond improves accordingly with increasing bonding pressure. The oxide layer at the bonding interface underwent an evolution process from continuous to discontinuous and ultimately disappeared with the increase of bonding pressure. The optimal joint was achieved in this work at the pressure of 20 MPa. Considering the research of reliability of joints in engineering applications, the optimum joint was subjected to post-bonding heat treatment (PBHT) of 750 °C for 90 minutes. The diffusion bonded sample subjected to PBHT exibits microstructural features and tensile properties similar to those of the base metal. It is noteworthy that the tensile fracture does not occur at the bonding interface, but in the base metal far away from the interface. This work can provide a feasible method for the removal of oxides at the diffusion bonding interface of RAFM steel and also contribute to the establishment and optimization of diffusion bonding processes, thereby improving the quality of the diffusion bonding joint.

Keywords: Bonding pressure; Oxide layer; Microstructure; Mechanical property

1. Introduction

Reduced activation ferritic/martensitic (RAFM) steel is frequently chosen as the primary material for the main structure of fusion reactor blanket module due to its impressive high-temperature mechanical attributes, superior radiation swelling resistance, low thermal expansion coefficient, and favorable processing performance [1-3]. Reliable welding methodology for RAFM steel is the key to manufacturing blanket module. The fusion welding involves the melting of the base metal to create a molten pool, which then solidifies through a nonequilibrium process. Thus, fusion welding techniques such as laser beam or arc welding or even solid-state friction stir welding may result in the formation of undesirable brittle phases in structural steels [4-6]. In addition to the undesirable variations in the microstructure and mechanical properties, fusion welding also lead to a concentration of residual stress in the welded joint, which often becomes the weakest link of components. More importantly, a large number of researches show that long-term creep in fusion welded joints of high Cr ferritic/martensitic steel leads to inevitable type IV fracture failure, resulting in the creep life of joints much lower than that of base metal [7-9]. The fact that the blanket module is continuously subjected to high-temperature and high-radiation conditions only further deteriorates the strength of the joint. Therefore, the solid phase welding technology represented by diffusion bonding is expected to make up for the shortcomings of fusion welding and become a potential alternatives to fusion welding for the assembly of RAFM steel components, due to its outstanding advantages such as low welding temperature, small welding deformation, and high dimensional assembly accuracy.

Diffusion bonding is a solid bonding process that applies pressure to two or more solid phase materials at high-temperature to cause microscopic plastic deformation at the bonding interface, and maintains a period of time to form a solid metallurgical bond through atomic diffusion [10]. The key elements influencing diffusion bonding include temperature, pressure, and time. So far, some research has been conducted on the diffusion bonding of RAFM steel



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[11-13]. However, most of these studies have primarily focused on the variables of bonding temperature and bonding time, with limited attention given to the bonding pressure. This article intends to explore the role of bonding pressure in the diffusion bonding of RAFM steel. Additionally, considering the practical engineering applications, the joint with the best bonding quality in this study was subjected to post bonding heat treatment (PBHT), and its mechanical properties were assessed.

2. Materials and experiment methods 2.1. *Diffusion bonding experiment*

The RAFM steel was cast in a vacuum induction furnace into an ingot weight 25 kg and then forged into cylinder bars with a size of 350 mm in height and 120 mm in diameter. The chemical composition of the experimental steel can be found in Table 1. The initial RAFM steel base sample was normalized at 1050 °C for 30 minutes and tempered at 750 °C for 90 minutes. Its microstructure reveals tempered martensite with a small amount of δ -ferrite, as depicted in Fig. 1. The substrate used for processing the diffusion bonded sample was cut out from the steel sample and uniformly polished, resulting in a disc with 60 mm diameter and 10 mm height. Then, cylindrical sample blocks with a diameter and height of 10 mm were cut from the disc. The blocks were paired up to conduct diffusion bonding test using the Gleeble-3500 thermal simulator. Prior to diffusion bonding, the sample

Table 1. Chemical composition of the steel (in wt.%)



Figure 1. OM image of the original microstructure of RAFM steel

blocks were ultrasonically cleaned in acetone and alcohol and then dried. The sample blocks to be bonded were heated to $1050 \,^{\circ}$ C at a heating rate of $100 \,^{\circ}$ C/min, and maintained under various pressure namely 10, 15, and 20 MPa for 60 minutes. After that, the pressure was relieved and then the bonded specimen was cooled to room temperature. The vacuum level during diffusion bonding was maintained at 1×10^4 Pa.

2.2. Microstructure observation and tensile property

The examination of the microstructure at diffusion bonding interface was carried out with a Zeiss Gemini SEM500 scanning electron microscope (SEM). Following grinding and mechanical polishing, the samples were treated with an etching solution composed of 20 mL hydrochloric acid, 100 mL water and 5g ferric chloride. The evolution of oxides at the diffusion bonding interface under varying bonding pressures was studied using an energy dispersive spectroscopy (EDS). The precipitates of diffusion bonded sample after PBHT were identified using a Jem-2100f transmission electron microscopy (TEM). For testing the tensile attributes of the diffusion bonding sample, a specimen with a 4 mm gauge length was employed. The bonding interface was located in the middle of the tensile specimen. The tensile properties were evaluated at room temperature on an Instron 8871 using a strain rate of 1×10^{-3} s⁻¹.

3. Results and discussion 3.1. Microstructure of the diffusion bonded samples

3.1.1. SEM observation

Fig. 2 presents the SEM images of the diffusion bonding interface obtained at 1050 °C for 60 minutes upon pressures of 10, 15, and 20 MPa, respectively. The bonding interface is indicated by the black arrow in the figure. It can be observed that there are no remarkable microstructure differences or gradients in close proximity to the bonding interface. Due to the fact that this work involves the direct diffusion bonding of metal solid, there is an absence of a transition layer at the bonding interface. The bonding interface of joint obtained at 10 MPa is easily identifiable, implying that the joint has not yet achieved complete metallurgical bonding. As the bonding pressure intensifies, the interface gradually becomes blurry. When the bonding pressure is 15 MPa, some grains on both sides of the bonding interface grow reciprocally and traverse the bonding interface. The mutual growth and fusion of grains on both sides of the bonding interface into new grains represents a good development trend in bonding





Figure 2. SEM images of RAFM steel diffusion bonded joints held at 1050 °C for 60 minutes upon different pressures: (a) 10 MPa, (b) 15 MPa, (c) 20 MPa

quality [14]. The bonding pressure is further increased to 20 MPa, the bonding interface is difficult to distinguish, and the microstructure on either side of the bonding interface is fully continuous. It can be concluded that increasing the bonding pressure from 10 to 20 MPa significantly enhances the quality of the diffusion bonded joint.

3.1.2. EDS analysis

Fig. 3 shows the EDS mapping analysis of the diffusion bonding interface obtained at 1050 °C for 60 minutes upon pressures of 10, 15, and 20 MPa, respectively. It reveals a high concentration of Chromium (Cr), Oxygen (O) and manganese (Mn), and a low concentration of Iron (Fe) around the bonding interface at a pressure of 10 MPa, and the distribution of Cr/Mn rich oxide is continuous, as displayed in Fig. 3(a). The thickness of the oxide layer is approximately 0.65 µm. It should be pointed out that the Cr content in RAFM steel is relatively high. Despite meticulous sanding and polishing were carried out during the preparation process of the samples to be bonded, even short exposure of the machined surface allows Cr to easily react with O, forming Cr rich oxides [15]. Meanwhile, the Mn content is also higher at the interface of 10 MPa and

15 MPa, as Mn is also easily bound to O. There is no significant difference in tungsten (W) concentration at the interface. As the bonding pressure escalates from 15 to 20 MPa, the concentration fluctuations of Cr, Mn, Fe and O at the interface disappear, and the oxide gradually becomes sporadic until it entirely vanishes, as illustrated in Figs. 3(a) and 3(c). This occurs because a higher bonding pressure augments the probability of plastic deformation of micro protrusions on the bonding surface, thereby increasing the close contact area at the bonding interface and leading to the rupture of the oxide layer. Moreover, Sharma et al. [16] confirmed that plastic deformation at the interface caused by pressure can bolster the plastic flow of the material at the bonding interface and facilitate the rupture of the oxide layer. The rupture of high-level oxide layer caused by high bonding pressure effectively heightens the direct contact of exposed metals, providing sufficient diffusion channels for atomic diffusion, which is more favorable for achieving mutual diffusion between atoms [17]. Simultaneously, the plastic deformation at the bonding interface places the atoms at that location to be in a state of high deformation energy, making them easier to diffuse [18]. In addition, EDS analysis also validates that as the bonding pressure increases from 10 to 20 MPa, the element distribution at the





Figure 3. EDS mapping analyses of RAFM steel diffusion bonded joints held at 1050 °C for 60 minutes upon different pressures: (a) 10 MPa, (b) 15 MPa, (c) 20 MPa

bonding interface is becoming increasingly uniform, and there is no fluctuation in element content. This further corroborates the adequacy of element diffusion at the interface. It should be emphasized that while bonding pressure can improve the quality of joints, excessive bonding pressure can cause deformation or even cracking of the workpiece, which is not beneficial for manufacturing structural components. The selection of bonding pressure should balance meeting engineering application requirements and preventing deformation of the welded parts.

3.1.3. SEM and TEM observations of the diffusion bonded sample subjected to PBHT

It is found that the bonding quality of the diffusion bonding sample produced at 1050 °C for 60 minutes upon the pressure of 20 MPa is the best in this study through prior analysis of the diffusion bonding



process. Thus, the diffusion bonded joint obtained at 20 MPa is selected as the subsequent research object. Given the requirements of engineering applications, the joint is subjected to a PBHT of 750 °C tempering for 90 minutes. Fig. 4 shows the SEM and TEM images of the diffusion bonding sample after PBHT. SEM detection reveals a significant amount of precipitates in the microstructure, while the bonding interface remains indistinct. The precipitates near the diffusion bonding interface were extracted by means of carbon extraction replica technique and examined by TEM. It was found that the precipitates are primarily located at the prior austenite grain boundary (green schematic line diagonally downwards to the right) and martensite lath boundary (yellow schematic line diagonally upward to the right), as illustrated in Fig. 4(b). The type of precipitate is determined as M₂₃C₆-type carbides by electron diffraction pattern analysis. This is similar to the precipitation strengthening method of RAFM steel without diffusion bonding after tempering treatment in previous studies, that is, $M_{23}C_6$ -type carbides are precipitated at the prior austenite grain boundary and martensite lath boundary to achieve precipitation strengthening [19, 20]. The above analysis indicates that there is no significant difference in microstructure and precipitation strengthening method between



Figure 4. SEM and TEM images of RAFM steel diffusion bonded joint held at 1050 °C for 60 minutes upon 20 MPa subjected to PBHT

RAFM steel with direct diffusion bonding followed by PBHT and the base metal without bonding after tempering treatment.

3.2. Mechanical property of the diffusion bonded sample

The tensile properties of the optimal bonded RAFM steel (obtained at 20 MPa) were tested before PBHT, with a tensile strength of 950 MPa and an elongation of 8.5%. The plasticity is too poor. Moreover, the final service state of RAFM steel is a high-temperature tempered state, which can ensure the precipitation of a large amount of precipitates in the matrix to meet all performance requirements [21]. Therefore, PBHT of the diffusion bonded joint is necessary. Fig. 5 presents the comparison between the tensile properties of the diffusion bonded sample after PBHT with the original normalized and tempered base sample. It can be found that the tensile strength and total elongation of the diffusion bonded sample are comparable to those of the original base metal sample, with a tensile strength around 680 MPa and a total elongation of approximately 22%. Observing the diffusion bonding sample after fracture, it is evident that the fracture position is located on the base metal far from the bonding interface, as shown in the tensile fracture pattern in the upper right corner of Fig. 5. This can be explained by the fact that the plastic deformation caused by micro protrusions at the bonding interface during diffusion bonding is relatively higher than that of the base material which leads to a higher dislocation density at the interface and more sub-grain boundaries are formed to precipitate more carbides during PBHT, thus resulting in higher strength at the bonding interface [22]. SEM examination on the tensile fracture of the diffusion bonded sample is performed, as demonstrated in Fig. 6. Significant necking can be observed in the tensile fracture observed at low magnification, accompanied by obvious shear lips at the outer periphery, as shown in Fig. 6(a). High magnification observation of tensile fracture surface in Fig. 6(b) reveals the presence of conspicuous dimples with various sizes in the fracture, indicating that the fracture mechanism is ductile fracture.

Based on the above analysis, it can be preliminarily considered that the diffusion bonding of RAFM steel can satisfy the requirements of engineering applications. This can be explained by the following two reasons. On the one hand, the microstructure and element transition of the prepared bonding joint are uniform, and there is no formation of non-equilibrium microstructure. On the other hand, the strength and plasticity of the bonding joint are not lower than those of the base metal.





Figure 5. Comparison of tensile property for original base metal and diffusion bonded joint (prepared at 1050 °C for 60 minutes upon 20 MPa) subjected to PBHT



Figure 6. SEM images at low (a) and high (b) magnification of tensile fracture of RAFM steel diffusion bonded joint held at 1050 °C for 60 minutes upon 20 MPa subjected to PBHT

4. Conclusions

This work investigated the influence of bonding pressure on the RAFM steel diffusion bonded joints, and the microstructure and mechanical property of the optimal joint subjected to PBHT. The main results are as follows:

(1) The quality of the diffusion bonded joints held at 1050 °C for 60 minutes gradually improved as the bonding pressure increased from 10 to 20 MPa. The specific manifestation was that the oxides at the bonding interface evolved from continuous distribution to discontinuous distribution, and ultimately disappeared completely.

(2) After PBHT, the tensile properties of the diffusion bonded joint obtained at 1050 °C for 60 minutes with a pressure of 20 MPa were comparable to those of the base metal.

(3) The tensile fracture of the diffusion bonded specimen was located at the base metal far from the bonding interface.

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Author contributions

Jianguo Chen: Experimental scheme and design, Paper writing, Supervision. Wanjun Wang: Investigation. Ji Dong: Translation and revision. Chuancai Wang: Investigation. Yushun Wei: Formal analysis.

Data availability

All data generated or analyzed during this work are included in this published article.

Conflicts of interests

The authors declare that they have no known competing financial interests that could have appeared to influence the work reported in this paper.

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DIFUZNO SPAJANJE RAFM ČELIKA: NASTANAK MEĐUFAZNOG SLOJA OKSIDA POD PRITISKOM I NJEGOVA MEHANIČKA SVOJSTVA NAKON TERMIČKE OBRADE

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Apstrakt

U ovom radu je ispitivan uticaj povećanja pritiska sa 10 na 20 MPa (5 MPa po intervalu) na 1050°C tokom 60 minuta na difuznom spoju RAFM čelika. Rezultati pokazuju da kvalitet spoja raste proporcionalno sa povećanjem pritiska. Sloj oksida na granici spoja prolazi kroz proces evolucije od kontinuiranog do diskontinuiranog i konačno nestaje sa povećanjem pritiska. Optimalni spoj je postignut pri pritisku od 20 MPa. Razmatrajući istraživanja pouzdanosti spojeva u inženjerskim primenama, optimalni spoj je podvrgnut termičkoj obradi (PBHT) nakon nastanka spoja na 750 °C tokom 90 minuta. Difuzni spoj podvrgnut PBHT pokazuje mikrostrukturne karakteristike i mehaničke osobine slične onima osnovnog metala. Važno je napomenuti da se lom pri zatezanju ne javlja na granici spoja, već u osnovnom metalu daleko od granice. Ovaj rad može pružiti izvodljivu metodu za uklanjanje oksida na granici difuznog spoja kod RAFM čelika, kao i doprineti uspostavljanju i optimizaciji procesa difuznog spajanja, čime se poboljšava kvalitet difuznog spoja.

Ključne reči: Pritisak prilikom spajanja; Sloj oksida; Mikrostruktura; Mehanička svojstva

